

# Download Solution Manual Levine Physical Chemistry

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Electronic structure. Singlet oxygen refers to one of two singlet electronic excited states. The two singlet states are denoted  $1^1\text{O}_2$  and  $1^3\text{O}_2$  (the preceding superscripted "1" indicates it as a singlet state). The singlet states of oxygen are 158 and 95 kilojoules per mole higher in energy than the triplet ground state of oxygen. Under most common laboratory conditions, the higher energy ...